Rotamers of 1-Phenylethylamine Ligands in (R, S)- and (S, S)-trans-Bis(succinimidato)-bis(1-phenylethylamine)palladium(II) Complexes

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Both trans-[Pd(succim)₂(R-phenea)(S-phenea)] and trans-[Pd(succim)₂(S-phenea)₂] (succim=succinimidate, phenea=1-phenylethylamine) were prepared and their crystal structures were determined by X-ray analysis. Three different staggered rotamers around N-C bond of the phenea ligands have been found.

The structural difference between *meso* and *optically active* forms of the metal complexes sometimes plays a crucial role in chemical reactions. Indeed, Noyori et al. recently reported those zinc complexes with chiral amino alcohols in relation to the chiral amplification reaction.¹⁾ We wish to report here the conformational change of 1-phenylethylamine ligands in *meso* and *optically active* palladium(II) complexes with monodentate succinimidate ligands.

Two new complexes trans-[Pd(succim)₂(R-phenea)(S-phenea)] 1 and trans-[Pd(succim)₂(S-phenea)₂] 2 were prepared following the reported method using the racemic or optically active phenea ligand.²⁾ Suitable crystals were obtained from chloroform/hexane (1:1) mixture, and X-ray structure determinations of 1 and 2 were performed.³⁾ ORTEP drawings of 1 and 2 are shown in Fig. 1.

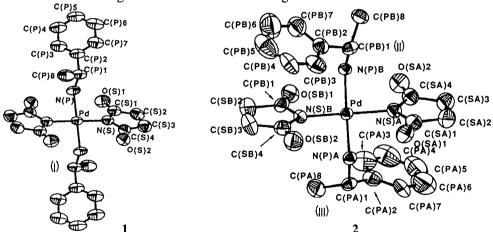


Fig. 1. View of the structures of 1 and 2. Selected bond lengths(Å) and angles (°): Pd-N(S) = 2.048(5), Pd-N(P) = 2.067(5), N(S)-Pd-N(P) = 89.7(2), for 1; Pd-N(S) = 2.077(9), Pd-N(P) = 2.060(8), Pd-N(S) = 2.013(8), Pd-N(P) = 2.042(8), Pd-N(S) = 2.

In 1, the palladium atom is located on a special position 1. The dihedral angle between the PdN₄ and the imidate CNC planes is 41°. Intramolecular hydrogen bonds between the phenea NH₂ and succim carbonyl oxygen are N-H...O 2.83(1) Å, N-H-O 141(4)° and 2.86(1) Å, 140(4)° respectively.

In 2, the geometry around the palladium atom is still nearly centrosymmetric except the methyl groups of S-phenea. Moreover, phenyl planes of both S-phenea are situated nearly parallel to succim planes. This arrangement forces two methyl groups to have different orientations. The corresponding dihedral angles are 42° and 45° . Intramolecular hydrogen bonds N-H...O 2.85(1) -2.94(1) Å, N-H-O 131(7) - $140(7)^{\circ}$ resemble those found in 1.

The most remarkable feature of the molecular structures of 1 and 2 is the presence of three different possible *staggered* rotamers of phenea. The projections of phenea along the N-C(chiral center) bond are shown in Fig. 2. Three types of the rotamer appear; those are *staggered*-(I) for 1, *staggered*-(II) and *staggered*-(III) for 2 respectively. (I) is so called opened form and (II) and (III) correspond to the closed form.⁴⁾ The Pd-N(phenea) distances suggest that the steric repulsions between phenea and succim in (I) and (III) are larger than that in (II).⁵⁾ Therefore, it can be mentioned that the repulsion caused by the methyl group is larger than that of the phenyl group.

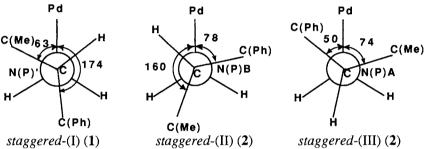


Fig. 2. Three staggered rotamers around N-C(chiral center) of phenea.

The investigations of the isomerization reaction in the solution state are now in progress. This work was partially supported by Grant-in-Aid for Scientific Research No. 03640534 (Y. K) from the Ministry of Education, Science and Culture.

References

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- 2) S. Komorita and S. Yamada, *Synth. React. Inorg. Met-org. Chem.*, **14**, 485 (1984). Elemental analysis: Found (Calcd for C₂₄H₃₀N₄O₄Pd): C, 53.03(52.90); H, 5.54(5.55); N, 10.32(10.28) % for **1**, and C, 52.17(52.90); H, 5.53(5.55); N, 10.10(10.28) % for **2**.
- 3) Crystal data of 1: F.W.=544.94, triclinic, $P\overline{1}$, a=11.478(2), b=8.483(3), c=7.129(1) Å $\alpha=102.15(2)$, $\beta=106.17(1)$, $\gamma=67.14(1)$ °, U=610.2(2) Å³, Z=1, Dc=1.48, Dm=1.47 g cm⁻³, $\mu(Cu K\alpha)=65.7$ cm⁻¹. R=0.056, Rw=0.065. The absorption effects were corrected. Total 1834 independent reflections with $|Fo|>3\sigma(Fo)$ (2θ max=125°) were used. Crystal data of 2: F.W.=544.94, monoclinic, $P2_1$, a=12.080(3), b=7.854(1), c=13.254(2) Å $\beta=99.77(2)$ °, U=1239.3(4) Å³, Z=2, Dc=1.46, Dm=1.45 g cm⁻³, $\mu(Mo K\alpha)=7.71$ cm⁻¹. R=0.033, Rw=0.066. Total 2434 independent reflections with $|Fo|>3\sigma(Fo)$ (2θ max=60°) were used.
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- 5) The averaged value of Pd-N (primary amine) is 2.041(22) Å: A. G. Orpen, L. Brammer, F. H. Allen, O. Kennard, D. G. Watson, and R. Taylor, *J. Chem. Soc.*, *Dalton Trans.*, **1989**, S36.

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